

FABRICACIÓN DEL PIGMENTO CERÁMICO $ZnCr_2O_4$ TIPO ESPINELA POR MÉTODOS NO CONVENCIONALES

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RESUMEN

En este trabajo se sintetizó el pigmento cerámico con estructura espinela $ZnCr_2O_4$ por los métodos no convencionales de autocombustión, coprecipitación asistida por ultrasonido, reacción en miscelas normales y molienda de alta energía. Los pigmentos obtenidos se caracterizaron por Difracción de Rayos X (DRX), con el fin de observar la formación de la estructura esperada y Colorimetría CIEL*a*b* para determinar las coordenadas colorimétricas. El objetivo de este trabajo fue comparar cuatro métodos alternativos y determinar cuál de estos métodos brindaba mejores resultados en cuanto a la estructura cristalina, (espinela) y las coordenadas de color, teniendo unas condiciones de tratamiento térmico inferiores a las que se utilizan tradicionalmente que son superiores a los 1100 °C. Los resultados mostraron que por las cuatro rutas se puede obtener este pigmento cerámico con la estructura cristalina deseada a temperaturas inferiores a los 700 °C, además de obtener unas tonalidades similares a la cromita de zinc utilizada empleada en la industria cerámica, las cuales varían de a verde a verde-grisáceo.

Anexo 53. Publicación del docente Juan Fernando Montoya Carvajal en la Revista Colombiana de Materiales, No. 5, 2014. p.284-289

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Magnetic response of polycrystalline $\text{YBaCo}_4\text{O}_{7-\delta}$ synthesized through the physical and chemical route: The role of phase inhomogeneities

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ABSTRACT

Polycrystalline $\text{YBaCo}_4\text{O}_{7-\delta}$ samples were obtained through a standard solid state reaction, and their structural, morphological, electrical, and magnetic properties are carefully studied. The X-ray powder diffraction (XRD) patterns showed reflections of a pure hexagonal structure (space group $P6_3mc$) with lattice parameters being very close to those reported in the literature. Although XRD analysis showed that the main phase present is $\text{H}4$, the presence of secondary phases could not be ruled out based solely on the XRD characterization. Indeed, sensitive SQUID magnetic measurements showed that the samples were affected by very small quantities of the $\text{H}2$ phase ($\text{YBaCo}_4\text{O}_{6.5}$), which typically manifests itself through a conspicuous increase in the magnetization at ~ 300 K. The results achieved corroborated the predictions concerning the difficulty of stabilizing the $\text{H}4$ phase when synthesized via the standard solid-state reaction. With this in mind, we next attempted to obtain the compound with improved phase purity. In so doing, the $\text{YBaCo}_4\text{O}_{7-\delta}$ compound was synthesized through a wet chemistry method based on a sol-gel route. The XRD patterns recorded for these samples revealed well-defined peaks corresponding to a pure hexagonal structure. More interestingly, SQUID measurements show no sign of features in the $M(T)$ curve at temperatures as low as ~ 80 K. This result was consistent with the magnetic behavior observed in $\text{YBaCo}_4\text{O}_{7-\delta}$ single-crystals. At temperatures below ~ 80 K, a clear feature was observed which seemed to correlate with a transition into an antiferromagnetic state. Isothermal magnetization recorded at 70 K showed that field-induced effects manifested themselves through the appearance of a ferromagnetic-like component. This ferromagnetic component may arise from spin canted of the underlying antiferromagnetic state or through field-induced structural transition (at least at local scale). Although a definitive interpretation of the in-field behavior from magnetization data alone is difficult because of the unknown role of the yttrium ion, the results achieved suggest that the magnetic behavior observed in members of the $\text{R}_2\text{BaCo}_4\text{O}_{7-\delta}$ family is not necessarily linked to the moment of the rare-earth ion, as in case of $\text{YBaCo}_4\text{O}_{7-\delta}$, since the yttrium ion is not magnetic. Beyond this important finding, the experimental results reported in the present paper demonstrate that the tested chemical route is suitable for synthesizing complex, single-phase oxides, such as the $\text{YBaCo}_4\text{O}_{7-\delta}$ cobaltate. The success in synthesizing high-purity $\text{YBaCo}_4\text{O}_{7-\delta}$ allows one to subtract parasitic effects from the intrinsic magnetic behavior of this challenging system.

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Anexo 54. Publicación del docente Juan Fernando Montoya en el Journal of Magnetism and Magnetic Materials, Vol. 360, 2014.

APLICACIÓN DEL MODELO DE COLOR KUBELKA - MUNK EN ESMALTES DE CERÁMICA TRADICIONAL

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RESUMEN

El objetivo de este trabajo consistió en aplicar del modelo de Kubelka Munk para la predicción de color en esmaltes utilizados en decoración de loza cerámica. El modelo relaciona los coeficientes de absorción y de dispersión de reflectancia difusa esperados para una mezcla de tres pigmentos cerámicos en el espectro visible con las respectivas concentraciones de la composición del esmalte, con lo cual es posible predecir un tono determinado. Se empleó la espectrofotometría visible con el fin de obtener los coeficientes de absorción y dispersión de los pigmentos elegidos sobre sustratos cerámicos para lojería, y también para los espectros de reflectancia difusa de las mezclas elegidas para los esmaltes calcinados, los resultados obtenidos mostraron una buena predicción en algunas regiones del espectro visible, concluyéndose que el modelo podría utilizarse para el control de calidad de los productos fabricados en la industria cerámica.

Palabras clave: Color, Kubelka Munk, Esmalte cerámico, Pigmento cerámico, Medición de color.

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APPLICATION OF THE KUBELKA - MUNK COLOR MODEL IN ENAMEL OF TRADITIONAL POTTERY

ABSTRACT

The objective of this work is to apply the Kubelka Munk model for predicting color glazes used in ceramic pottery decoration. The model relates the absorption coefficients of scattering and diffuse reflectance expected for a mixture of three ceramic pigments in the respective concentrations of the composition of the enamel, which predict a certain tone. It was used visible spectrophotometry to obtain the absorption and scattering coefficients of pigments chosen on ceramic substrates for their use in pottery, and also for the diffuse reflectance spectra of mixtures chosen for enamels calcined, the results showed a good prediction in some regions of the visible spectrum, concluding that the model could be used for quality control of products manufactured in the ceramic industry.

Keywords: Color, Kubelka Munk, Ceramic enamel, Ceramic pigment, Color measurement.

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Microencapsulation of Banana Passion Fruit (*Passiflora tripartita* Var. *Mollissima*): A New Alternative as a Natural Additive as Antioxidant

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Abstract

Banana passion fruit (*P. tripartita* var. *Mollissima*) is one of the most promising tropical fruits giving its antioxidant activity (AOA) to replace synthetic additives. Despite this property, there are no studies about the metabolites responsible for its biological function or proposals for the application of technologies, such as microencapsulation by spray drying, to improve its properties and ease its incorporation in several food matrices. The aim of this study is to microencapsulate the pulp of banana passion fruit with several mixtures of encapsulants and identify which one of these mixtures is better to preserve its AOA. The antioxidant activity values for the banana passion fruit pulp were as follows: DPPH: 6630.2 ± 91 μ Mtrolox/100g; ABTS: 18764.3 ± 270.4 μ Mtrolox/100g; FRAP: 1703.6 ± 938.2 mgAA/100g. ORAC: 8105.4 ± 424.2 μ mol TEAC/100g of sample; Total phenols: 8862.2 ± 451.4 gallic ac. mg/100g. The concentrations of the bioactive compounds expressed in mg of gallic acid per 100 g of the pulp on a dry base were 13.9 ± 0.004 ; 5.9 ± 0.001 and 126.3 ± 0.004 for caffeic, p-coumaric and ferulic acids, respectively. The best shelf-life followed by ABTS in eight assays was between 28.8 and 31.5 weeks using maltodextrin and modified starch, MD:MS (1/4:3/4) and MD:MS (0:1), respectively. In conclusion, ABTS is the best method to measure the AOA in banana passion fruit because it correlated with the phenolic compounds better than DPPH and FRAP methods. Additionally, two options were found to protect the AOA and to extend the shelf-life of the passion fruit by spray-drying, with mixtures of encapsulants widely used in the food industry.

Anexo 56. Publicación de los docente Leonidas de Jesús Millán Cardona, Ana María Restrepo Duque, Luz María Alzate Tamayo y Maritza Andrea Gil Garzón en la revista Food and Nutrition Sciences, No.5, 2014.

Carbon sequestration in different wetland plant communities in the Big Cypress Swamp region of southwest Florida

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Wetlands offer many ecosystem services, including the long-term sequestering of carbon (C) in soil. Here we present a study of C sequestration rates in a relatively undisturbed wetland landscape of southwest Florida. Accordingly, carbon sequestration was determined in four wetland plant communities and an adjacent hydric pine flatwood community that represent a gradient of inundation extent. Going from the wettest to the driest, communities were designated as: deep slough, bald cypress (*Taxodium distichum*), wet prairie and pond cypress (*Taxodium distichum* var. *imbricarium*). An adjacent hydric pine flatwood community was also included in the study as a reference upland site. Three soil cores were collected from each of these communities and were analyzed for total C content. Core samples were also analyzed for ¹³⁷Cs and ²¹⁰Pb activity to estimate accretion rates. C sequestration rates ($\text{g-C m}^{-2} \text{yr}^{-1}$) were the highest in the deep slough (98 ± 9) and bald cypress (98 ± 5) followed by the pond cypress (64 ± 7), wet prairie (39 ± 1) and pine flatwood (22 ± 5). These results suggest that impediment of decomposition by anaerobic conditions caused by prolonged wet cycles, may not account for all the variability in C sequestration rates observed in this subtropical setting. Instead, this variability could also be attributed to other factors like the quantity and chemical composition of the organic material reaching the soil. When methane emissions are taken into account, cypress-dominated (bald and pond cypress) and the deep slough communities act as net carbon sinks.

Keywords: carbon sequestration; subtropical wetland; cypress swamp; Everglades; climate regulation; *Taxodium distichum*

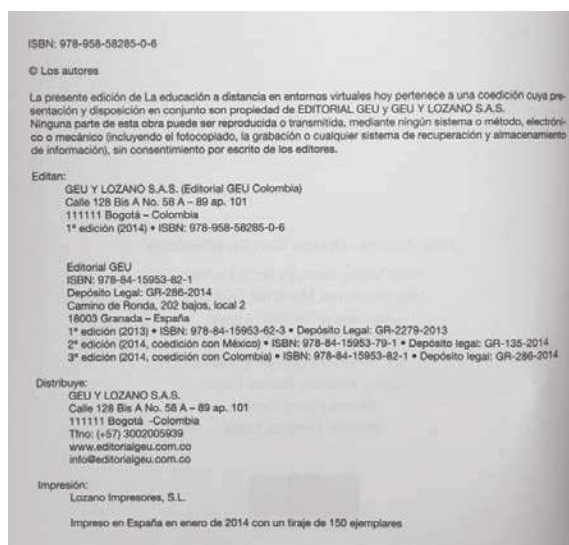
Introduction

Climate regulation through carbon (C) sequestration in wetlands soils may be one of the most important ecosystem services of wetlands in the long term. Carbon is sequestered in wetlands when C inputs (productivity and/or sedimentation) surpasses C outputs (decomposition and C exports) and the remaining organic material, mostly senesced plant material, is accumulated in the wetland's anaerobic sediment layer as a mat of partially decayed organic material, or peat. A fraction of this organic matter may also be incorporated into the mineral fraction of the soil as soil organic C. Globally, it is estimated that 455–700 Pg-C ($1 \text{ Pg} = 10^{15} \text{ g}$) of carbon in organic form is stored in wetlands (Mitsch & Gosselink 2015). By comparison, Lal (2008) estimates that there is 1550 Pg-C stored in the earth's soil organic C pool. This pool includes various forms of organic C, from highly active humus to relatively inert charcoal (Lal 2008). Considering that wetlands occupy only 5–8% of the terrestrial land surface (Mitsch & Gosselink 2015), these global estimates rank them as the terrestrial ecosystems with the highest C density (Kayranli et al. 2010), leading scientists from different disciplines to emphasize the key role that wetland ecosystems may play in the Earth's radiative forcing despite their relatively low percent coverage of the

terrestrial world (Whiting & Chanton 2001; Frohling et al. 2006; Page et al. 2011; Mitsch et al. 2013).

Values for C fluxes in wetlands, however, are far from definitive (Roulet 2000). For instance, early estimates of C sequestration rates for North America ($52.7 \text{ Tg-C yr}^{-1}$) and the world (137 Tg-C yr^{-1}) had an uncertainty of more than 100% according to Bridgman et al. (2006). More recently, Mitsch et al. (2013) revised this number and after including a revised area for tropical wetlands and their sequestration rates, they estimated that the worldwide sequestration may be around 830 Tg-C yr^{-1} . Much of our current knowledge of wetlands as carbon sources and sinks comes from extensively studied northern peatlands (Gorham 1991; Maltby & Immerzi 1993). In general, wetlands in boreal and subarctic biomes experience low temperatures that are partially responsible for inhibit organic matter decomposition and also limit productivity (Clymo 1984; Roulet et al. 2007). In the case of wetlands in warm subtropical and tropical climates, this temperature effect seems to be more complex and less understood. On the one hand, wetlands are generally more productive in lower latitudes. On the other hand, higher temperatures in these regions could lead to a rate of decomposition that exceeds that of productivity (Franzhuubbers et al. 2001; Mitsch et al. 2010).

Anexo 57. Publicación del docente Jorge Andrés Villa Betancur en International Journal of Biodiversity Science, Ecosystem Services & Management.



Anexo 58. Capítulo de libro publicado por la docente Lina María Rendón López

Guía estratégica de Propiedad Intelectual Universidad Empresa



Capítulo 7

Manejo de información confidencial

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Experto asesor de contenido
Juan Francisco Ortega Díaz²

7.1 INTRODUCCIÓN

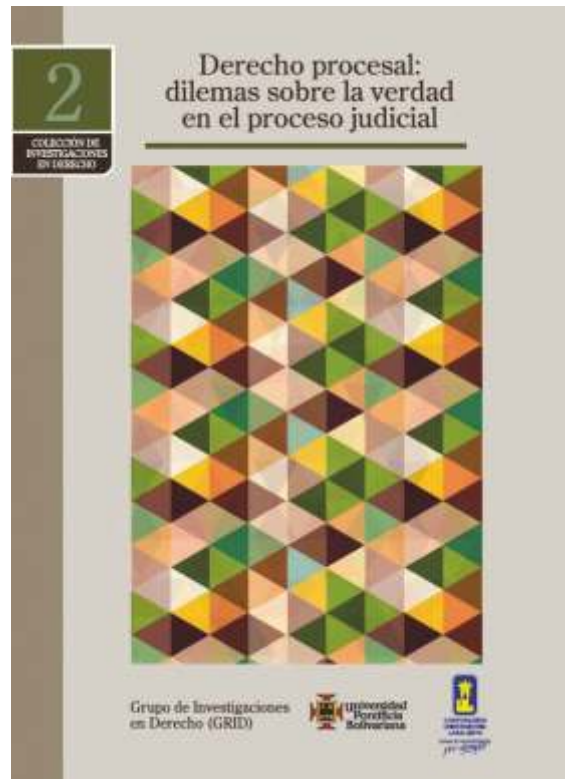
Como se ha demostrado a lo largo de la presente Guía, los proyectos de I+D generan nuevos conocimientos aplicados a la industria o al sector servicios, orientados a brindar soluciones o mejorar la forma en que se abordan los problemas técnicos de cualquier sector económico. La calidad, pertinencia y efectividad de los resultados obtenidos, aliados a la forma en que se gestionan, jurídica, administrativa y comercialmente, inciden directamente en que los mismos se clasifiquen como “innovadores”.

Buena parte de los resultados obtenidos en este tipo de proyectos pueden ser protegidos mediante diversas figuras jurídicas establecidas en las legislaciones de Propiedad Intelectual (PI) de cada país. En ocasiones, el secreto industrial puede reportar importantes ventajas al momento de iniciar procesos de protección, explotación y transferencia tecnológica toda vez que, a partir de su adecuada implementación, se permite documentar el conocimiento generado, definir el alcance y

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³ Docente en Derecho por la Universidad de Salamanca y autor de libros de texto de temáticas jurídicas consultivas en revistas internacionales. Profesor de Honor de la Universidad de los Andes y Director del Grupo de Estudios de Derecho de la Competencia y de la Propiedad Intelectual.



La conciliación. Su verdad como consenso en Colombia

Conciliation. Its truth as consensus in Colombia

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Resumen

La conciliación en derecho constituye una herramienta primordial para la terminación consensuada de los conflictos. Su finalidad es lograr paz entre las partes con la resolución

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Anexo 62. Capítulo de libro publicado por la docente Adriana Patricia Arboleda

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